Accelerator Mass Spectrometry Technique, and its applications in climate research

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AMS has been developed since the late 1970's

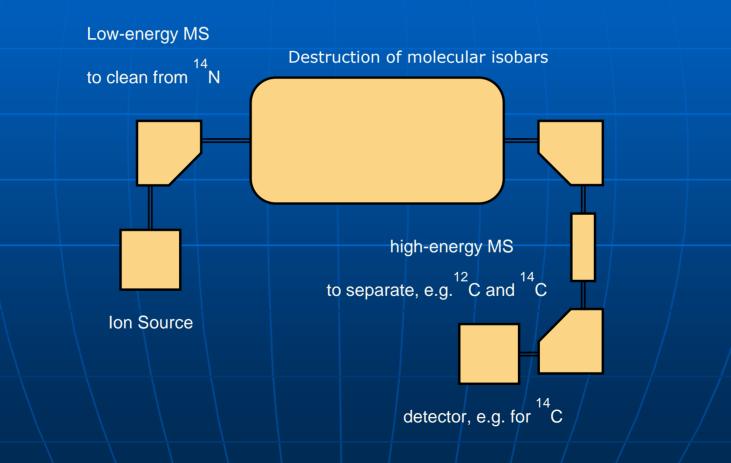
 Mass spectrometry rather than decay counting gains a sample mass reduction of ~10³

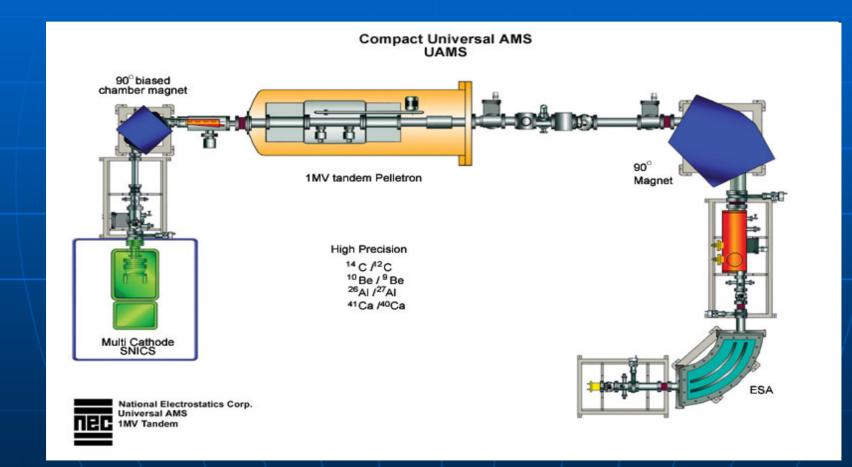
 Isotopes of interest ¹⁴C, ¹⁰Be, ²⁶Al, ³⁶Cl, ⁴¹Ca, ¹²⁹I



Brief description of the technique

General layout AMS accelerator





From big systems (>6MV) to very small ones: 250kV



 Measurements are expressed as ratios of the rare isotope (¹⁴C in this case) against the more abundant isotope (¹³C or ¹²C in this case)

¹⁴C / ¹³C

Samples are measured against a standard

Relative 14C content ("Fraction modern") $F = {}^{14}C/{}^{13}C_{sample} / {}^{14}C/{}^{13}C_{standard}$

Needs correction for isotopic fractionation

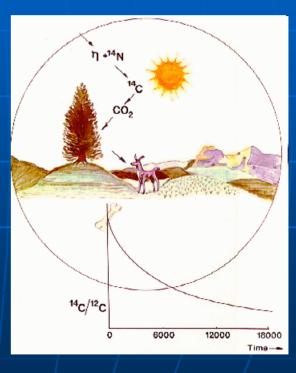
 Expression of ¹⁴C content
Scale for modern (contemporary) matter expressed as Δ¹⁴C in ‰

 0‰ is contemporary standard
-1000‰ is radiocarbon "dead" = no measurable radiocarbon content

¹⁴C and climate research

All living matter: "Modern" level 10⁻¹²

All organic matter >70KA: "Dead" level 10⁻¹⁵



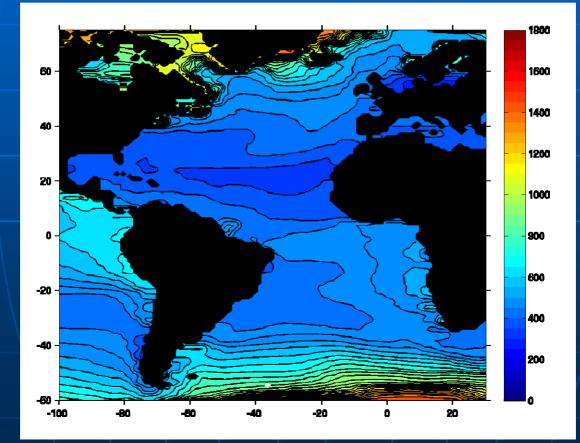
Design REM Hedges, Oxford

Carbon Sinks

- Atmosphere
- Ocean surfaceDeep ocean
- Biosphere
- Soil
- Rivers and lakes
- Fossil organic carbon
- Carbonates

- Different ¹⁴C content based on residence time of carbon and rate of exchange
- Measurement of ¹⁴C gives information about the dynamics of the system:
- Intake and uptake
- ¹⁴C time series
- Combined with [CO₂], quantitative apportionment of fossil contributions to atmosphere

Ocean radiocarbon reservoir: "apparent ¹⁴C ages"



R.G.Fairbanks

 ¹⁴C concentration in the atmosphere has changed as a result of fossil fuel additions and of the nuclear bomb tests

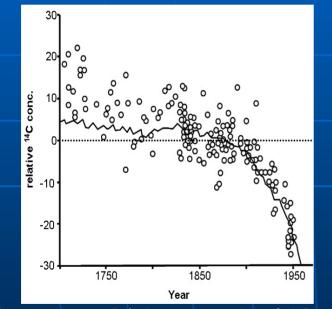
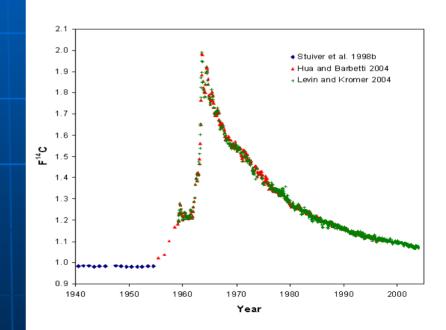


Figure 12.10: Relative concentration of 14C values of atmospheric CO2 derived from tree ring analysis. Trend line illustrates the dilution of 14C reflecting the burning of fossil fuels associated with world industrial activity known as the Suess Effect. From Molles (1992).





 This means we can source, or apportion the contributions of biogenic and fossil fuel-derived matter

 Because of relatively small sample size, we can measure specific extracts or specific components from larger molecules



Mandalakis et al. Data comparing Sweden, Croatia, Greece

TABLE 1. Yield, Purity, and Isotopic Composition of PAHs Isolated from Four Aerosol Samples

sampling sitea	yield (µg)	purity (%)₀	NOSAMS accession no.c	δ13 C (‰) d	∆14 C (‰) e	FBiomass(%) f
Aspvreten, Sweden (1995-1997)	14.1	95.4	OS-41865		-381	51
Aspvreten, Sweden (1998-2001)	35.7	96.1	OS-41862	-27.7	-388	50
Zadar & Velebit, Croatia (2003)	68.0	89.2	OS-43143	-29.2	-888	9
Finokalia, Greece (2003)	36.1	93.5	OS-43150	-29.0	-914	7

a The numbers in parentheses indicate the time period of aerosol sampling. b This purity was assessed prior to shipment from Stockholm University, and the subsequent additional cleanup at WHOI may have further increased these purities. c AMS accession numbers for each 14C analysis. d Standard deviation for all ä13C measurements is (0.1‰, based on replicate analysis of standards. e The relative standard error for these 14C data is 1-3%. (FBiomass is the percentage contribution of biomass burning to atmospheric PAHs calculated by an isotopic mass balance approach

 Extraction of specific PAHs from dust PM10 Aerosols using preparative capillary gas chromatography

Comparison of urban areas in Tokyo

 Seasonal variability, and differences between fractions for the importance of biomass burning: More important than expected

 Kumata et al: Compound specific 14C analysis of Polycyclic Aromatic Hydrocarbons associated with PM10 and PM1.1 Aerosols from Residential Areas of Suburban Tokyo. 2006

 AMS is a powerful, and thus essential tool to study anthropogenic climatic influences

 South-eastern Europe can expect more erratic climatic change than Northern Europe, so we need to develop a regional research infrastructure